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optic fibers used for visible and near-infrared light. Calculate the wavelength of one of these X-ray photons.

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represents an electron in or from an atomic orbital represents an electron that is physically identical to an electron in or from an atomic orbital, but this electron comes from the decay of a neutron to a proton and an electron it is also called a beta particle or ray. p.673.

21.1



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21.4.IDENTIFY: Use the mass  $m$  of the ring and the atomic mass  $M$  of gold to calculate the number of gold atoms. Each atom has 79 protons and an equal

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number of electrons. SET UP: 23 NA 6.02  
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