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CHAPTER SEVENTEEN

ELECTROCHEMISTRY For Review 1.

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Electrochemistry is the study of the interchange of chemical and electrical energy. A redox (oxidation-reduction) reaction is a reaction in which one or more electrons are transferred. In a galvanic cell, a spontaneous redox reaction occurs which produces an electric current. In an

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Galvanic Cell 17.3 Cell Potential,
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Section 18.1 Balancing

Oxidation-Reduction Equations

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18.2 - Balancing Oxidation ...

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Chapter 17 Electrochemistry Outline 1.

Voltaic cells 2. Standard voltages 3.

Relations between E° , ΔG° and K 4.

Electrolytic cells 5. Commercial cells

Electrochemistry • Electrochemistry is the study of the conversion of electrical and chemical energy • The conversion takes place in an electrochemical

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Chapter 17 [PDF]

Chapter Seventeen - Electrochemistry:

Animations and Documents:

Electrochemical Cell Part 1 This

animation shows the components of a zinc-copper cell. Electrochemical Cell

Part 2 This animation shows the lost of electrons in the zinc half-cell.

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