

O Level Chemistry Notes

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February 9, 2016 Igcse Chemistry Revision Notes, O Level Chemistry Revision Notes 1 Sulphuric Acid: It is a colourless oily yellow liquid, with density slightly higher than water. Having a boiling point of 338°C. It is soluble in water and emits heat when dissolved in water - exothermic in nature.

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$2\text{H}_2\text{O} + 4\text{e}^- \rightarrow 4\text{OH}^-$ - If we combine the ionic equations, we still get water as product of hydrogen and oxygen, but the energy produced is now electrical energy: $2\text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2\text{H}_2\text{O}(\text{g}) + \text{electrical energy}$

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in order, from lowest energy level to highest energy level • All the elements in a group have the same number of electrons in their outer ... O . O . H H + . • • • • ...

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$C_6H_{12}O_6(aq) \rightarrow 2C_2H_5OH(aq) + 2CO_2(g)$ Temperature is kept constant at 37 °C to prevent destruction of yeast at higher temperatures. Oxygen is removed by limewater and carbon dioxide is produced during fermentation. Alcohol is separated from solution by fractional distillation.
2. Reacting Ethene with Steam

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